

VOLUME 146

Rapid Thermal Annealing/ Chemical Vapor Deposition and Integrated Processing

EDITORS David Hodul Jeffrey C. Gelpey Martin L. Green Thomas E. Seidel



Rapid Thermal Annealing/Chemical Vapor Deposition and Integrated Processing

Symposium held April 25-28, 1989, San Diego, California, U.S.A.

EDITORS:

David Hodul Varian Research Center, Palo Alto, California, U.S.A.

Jeffrey C. Gelpey Peak Systems, Inc., Saugus, Massachusetts, U.S.A.

Martin L. Green AT & T Bell Laboratories, Murray Hill, New Jersey, U.S.A.

Thomas E. Seidel Sematech, Austin, Texas, U.S.A.



MATERIALS RESEARCH SOCIETY Pittsburgh, Pennsylvania

41

GROWTH AND DOPING KINETICS OF GexSi_{1-x} STRUCTURES BY LIMITED REACTION PROCESSING

P.M.Garone, J.C.Sturm, and P.V.Schwartz. Dept. of Electrical Eng. Princeton, NJ 08544

S.A.Schwarz and B.Wilkens , Bell Comunications Research Red Bank, NJ

ABSTRACT

We have investigated the growth rate and boron doping of $Si_{1-x}Ge_x$ epitaxial films grown by Limited Reaction Processing The growth experiments were carried out at a pressure of 6.0 torr with growth temperatures ranging from 625°C to 1000°C. The growth rate increases rapidly upon the additon of a small germane flow to the dichlorosilane in the reaction-rate-limited growth regime, and can not be explained simply by germanium incorporation. The presence of germane can increase the silicon growth rate by up to a factor of one hundred. Boron doping was also studied at high concentrations of boron in Si and Si_{1-x}Ge_x epitaxial films as a function of diborane flow and growth rate supports a simple kinetic model rather than an equilibrium model.

INTRODUCTION

Strained $Si_{1-x}Ge_x$ has been investigated over the past several years for application of heterojunctions to Si based device structures[1]. One successful method for growing high quality $Si_{1-x}Ge_x$ is LRP (Limited Reaction Processing)[2]. The growth and doping kinetics of these epitaxial films is of interest from both a scientific and technological (device structures) point of view.

LRP is a novel combination of chemical vapor deposition (CVD) and rapid thermal processing (RTP). The growth chamber consists of a round quartz tube in which a 4" wafer is supported, without a susceptor, on 3 quartz pins. The wafer temperature is adjusted by a bank of microprocessor controlled tungsten-halogen lamps, and the system is surrounded by a water-cooled gold-plated reflector.

The Si_{1-x}Ge_x layers are grown at 6.0 torr using 26 sccm dichlorosilane (DCS) in 3.0 slpm H₂ carrier gas along with varying flows of germane (GeH₄) and diborane (B₂H₆). Shown in figure 1 is the growth rate of Si (no GeH₄ flow) under these conditions from 700°C to 1000°C. One clearly notices the mass flow limited regime (above 800°C) and the reaction rate limited regime (below 800°C). In this low temperature regime the growth rate is determined by the actual chemical reaction rate. These low temperatures are also desirable to minimize interdiffusion between layers. This paper will focus on work in this regime. All work was done on Si <100> substrates.

Growth rates were calculated using layer thicknesses measured by Rutherford Backscattering or Secondary Ion Mass Spectroscopy(SIMS). Boron doping was determined with SIMS using implant standards.

DISCUSSION AND RESULTS

GexSi1-x Growth

As small amounts of GeH₄ (1-2 sccm compared to 26 sccm DCS) are added to the gas flow a marked enhancement is seen in the growth rate of epitaxial layers at 700°C (figure 2). With Si alone the growth rate is 30Å /min but it increases to over 300Å /min upon the addition of 2 sccm of GeH₄ to the 26 sccm of DCS. Two sccm of GeH₄ yields a film with composition Si_{0.8}Ge_{0.2}This growth rate enhancement may not simply be accounted for by considering the additional germanium growth. When the total growth rate is broken down into Ge and Si components (figure 2) we see that the Ge growth rate is linear with GeH₄ flow following simple first order growth kinetics. However the Si growth rate is tremendously enhanced by the GeH₄ up to 10 times the growth rate without GeH₄.

Growth at 700°C yielded epitaxial films, but "island" (3-dimensional) growth resulted in rough surfaces. Growth at 625° C however yielded surfaces that looked perfectly smooth when viewed via Nomarski microscopy. At 625° C we see similar results (figure 3) except that the Si growth rate enhancement is even more dramatic (up to 100x). The Ge component growth rate is virtually identical to the Ge component growth rate at 700°C and is also linear with GeH₄ flow. Since the Si growth rate at 625°C is substantially lower (even for GeH₄ enhanced growth) this leads to an increased Ge concentration in the epitaxial layer for a given GeH₄ flow. Similar results have been seen using SiH₄ by Uram et. al. at IBM [3].

Some films were grown at 900°C where mass transport, rather than reaction rate, determines the growth rate in our reactor. As expected the Si growth rate in the mass flow limited growth regime is unaffected by GeH_4 flow.

One possible approach to explaining the large Si growth rate increase is that the silicon surface bonds are mostly terminated by hydrogen, and that competition between hydrogen and the growth species for the available bonding sites limits the reaction rate. Because the Ge-H bond is weaker than the Si-H bond, the presence of Ge in the film would lead to faster hydrogen desorption, more open sites, and faster growth. If this were indeed true, the growth rate of pure Si (no GeH₄) should be strongly dependent on the hydrogen partial pressure and/or the DCS/H₂ ratio. Controlled experiments were carried out at 700°C varying the DCS/H₂ ratio (see Table I). The results show a negligable change in the growth rate while the DCS/H₂ ratio was changed by a factor of six. This suggests that hydrogen-desorption is not a limit to Si growth rate and not involved in the enhanced Si growth as Ge enters the epitaxial film.

Boron Dopant Incorporation

The boron dopant incorpoation as a function of diborane flow (11ppm in H_2) was investigated for Si at 700°C and 800°C. As may be seen in figure 4 the boron doping was linear with B_2H_6 flow. We see that this linearity extended above the solid solubility for 700°C silicon (2e19 cm-3) [4] suggesting a kinetic dependence for the boron incorporation rather than an equilibrium dependence. Equilibrium growth would set an upper limit to the doping levels (solid solubility).







Table I Effect of H2 Partial Pressure on 700ºC Si Growth Rates

H2 partial P (torr)	Growth Rate (Â/min)
5.95	24.2
17.95	27.4
1.95	21.5
	H2 partial P (torr) 5.95 17.95 1.95









A simple kinetic model for dopant incorporation would suggest that for a fixed boron (B_2H_6) flux impinging on the substrate an increased film growth rate would lead to a decreased boron density in the growing epitaxial film. Thus the final doping density of the film should be inversely proportional to the film growth rate. Comparing 800°C Si epi relative to 700°C epi we observe that the 10x increase in growth rate is matched by a 10x decrease in boron incorporation. This is further evidence for a simple kinetic model. Similar results for Ge_XSi_{1-X} films grown at 625°C have also been observed. Further evidence of the inverse relationship between concentration and growth rate was demonstrated by changing the GeH₄ flow with DCS flow fixed. As shown earlier, this significantly increases the growth rate. As seen in figure 5 as the growth rate increases, boron concentration decreases.

CONCLUSIONS

In the reaction rate limited growth regime the addition of small amounts of germane flow greatly enhances the overall growth rate of the film by catalyzing the Si growth rate, not simply by additional Ge growth. This enhancement is a large effect (10x to 100x depending on growth temp) and is more pronounced at lower temperatures. The mechanism for this enhancement is not known, however. The necessity of low temperatures for Si_{1-x}Ge_x (to avoid islanding) coupled with the high temperature requirements for pure Si (T \geq 700°C for reasonable growth rate) imply that rapid temperature switching of the substrate is necessary for the growth of multilayer structures.

Boron doping seems to follow a simple kinetic model rather than an equilibrium model in the reaction rate limited growth regime for both Si and Ge_xSi_{1-x} epitaxial films.

ACKNOWLEGEMENTS

This work was funded by NSF, ONR, and the NJ Commission on Science and Technology.

Special Thanks to S.A.Schwarz for SIMS analysis which made this work possible.

REFERENCES

1. J.C.Bean, T.T.Sheng, L.C.Feldman, A.T.Fiory and R.T.Lynch, Appl. Phys. Lett. 44, 102 (1984).

- J.F.Gibbons, C.M.Gronet and K.E.Williams, Appl. Phys. Lett. 47, 721 (1985).
- 3. Uram, Myerson, et. al., Electronic Materials Conference, Boulder CO. 1988.
- 4. S.K.Ghandi, VLSI Fabrication Principles, p. 70, Wiley-Interscience, 1983.